

2-(2-Oxopyrrolidin-1-yl)butyramide

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Key indicators

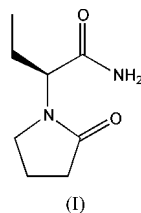
Single-crystal X-ray study
 $T = 293\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.007\text{ \AA}$
 R factor = 0.045
 wR factor = 0.144
Data-to-parameter ratio = 8.7

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound, $\text{C}_8\text{H}_{14}\text{N}_2\text{O}_2$, also known as levetiracetam and $(-)-(S)\text{-}\alpha\text{-ethyl-2-oxo-1-pyrrolidineacetamide}$, was synthesized from L-methionine. The pyrrolidinone ring has a half-chair conformation. Screw-related molecules are linked by $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds to form layers parallel to the bc plane.

Comment

Levetiracetam is an ethyl analog of the nootropic drug piracetam. It is a new antiepileptic drug (AED) (Bialer *et al.*, 1999), recently approved by the US Food and Drug Administration. It possesses a chiral center but only the S enantiomer has anticonvulsant activity, and therefore it is administered as a single enantiomer (Haria *et al.*, 1997). We report here the crystal structure of Levetiracetam, (I). The molecular structure of (I) is illustrated in Fig. 1. The structure reported and the scheme assume the S configuration for chiral C atom C6. The pyrrolidinone ring adopts a half-chair conformation. Weak $\text{N}-\text{H}\cdots\text{N}$, $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{N}$ interactions are observed in the molecular structure. In the crystal structure, screw-related molecules are linked by $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds to form layers parallel to the bc plane (Fig. 2 and Table 2).



Experimental

The title compound was prepared according to a previously published method (Cossement *et al.*, 1990) from L-methionine. L-Methionine was esterified to its methyl ester using absolute methanol and thionyl chloride. The resulting L-methionine methyl ester hydrochloride was amidated using gaseous ammonia to obtain L-methionine amide. This amide (50 mmol) was treated with potassium hydroxide (14.0 g), tetrabutylammonium bromide (2.5 mmol) and 4-chlorobutyl chloride (55 mmol) in dichloromethane to obtain $(S)\text{-}\alpha\text{-[2-(methylthio)ethyl]-2-oxopyrrolidineacetamide}$. This compound was reacted with Raney Nickel (T-1) to yield $(S)\text{-}\alpha\text{-ethyl-2-oxo-1-pyrrolidineacetamide}$. The reaction mixture was filtered and the filtrate was evaporated under reduced pressure. The residue was purified by column chromatography over silica (eluant: dichloromethane/methanol/ammonia, 94.5:4.3:0.2 $v/v/v$). The resulting powder was dissolved in 100 ml ethyl acetate. Single crystals of the title compound, suitable for X-ray analysis, were grown by slow evaporation of the solvent. $[\alpha]_{D_{25}} -87.5^\circ$ ($c = 1$ in acetone); $^1\text{H NMR}$

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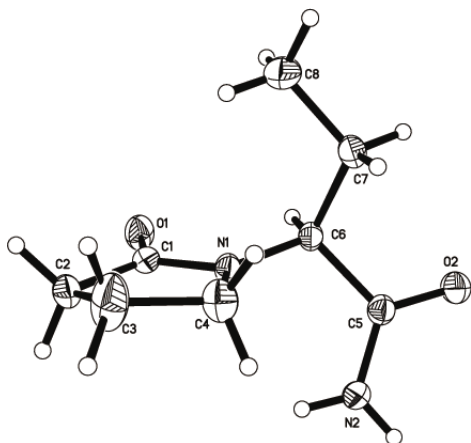


Figure 1
The molecular structure of (I), showing 30% probability displacement ellipsoids and the atom-numbering scheme.

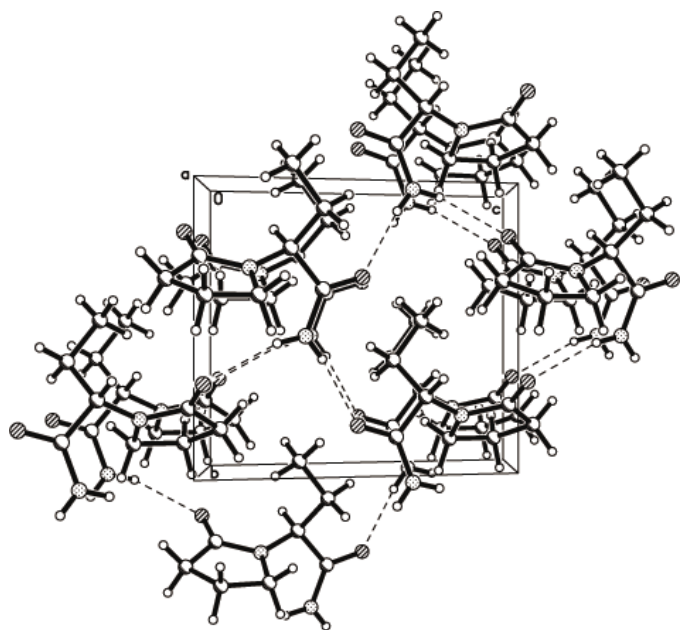


Figure 2
The crystal structure of (I), viewed along the *a* axis.

(CDCl₃, p.p.m.): 0.91 (2H, *t*), 1.66 (2H, *m*), 1.93 (2H, *m*), 2.36 (2H, *m*), 3.40 (2H, *m*), 4.45 (1H, *t*), 6.28 (2H, *d*); ¹³C NMR: 10.7 (CH₃), 18.3 (CH₂), 21.2 (CH₂), 31.3 (CH₂), 44.0 (CH₂), 56.3 (CH).

Crystal data

C₈H₁₄N₂O₂
M_r = 170.21
 Monoclinic, *P*2₁
a = 6.272 (5) Å
b = 7.993 (5) Å
c = 9.199 (7) Å
 β = 108.645 (9)°
V = 437.0 (6) Å³
Z = 2

D_x = 1.294 Mg m⁻³
 Mo *K*α radiation
 Cell parameters from 789 reflections
 θ = 3.4–23.1°
 μ = 0.09 mm⁻¹
T = 293 (2) K
 Plate, colorless
 0.20 × 0.18 × 0.16 mm

Data collection

Bruker SMART CCD area-detector diffractometer	731 reflections with $I > 2\sigma(I)$
ω scans	$R_{\text{int}} = 0.048$
Absorption correction: none	$\theta_{\text{max}} = 26.5^\circ$
2515 measured reflections	$h = -7 \rightarrow 7$
960 independent reflections	$k = -9 \rightarrow 9$
	$l = -4 \rightarrow 11$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0884P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.045$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.144$	$(\Delta/\sigma)_{\text{max}} = 0.001$
$S = 1.08$	$\Delta\rho_{\text{max}} = 0.23 \text{ e } \text{Å}^{-3}$
960 reflections	$\Delta\rho_{\text{min}} = -0.22 \text{ e } \text{Å}^{-3}$
110 parameters	Extinction correction: <i>SHELXL97</i>
H-atom parameters constrained	Extinction coefficient: 0.24 (4)

Table 1

Selected geometric parameters (Å, °).

O1—C1	1.231 (5)	C1—C2	1.507 (6)
O2—C5	1.231 (5)	C2—C3	1.500 (7)
N1—C1	1.352 (5)	C3—C4	1.484 (6)
N1—C6	1.451 (4)	C5—C6	1.538 (5)
N1—C4	1.473 (5)	C6—C7	1.521 (6)
N2—C5	1.325 (5)	C7—C8	1.508 (6)
<hr/>			
C1—N1—C6	122.9 (3)	C6—N1—C4	123.5 (3)
C1—N1—C4	112.7 (3)		

Table 2

Hydrogen-bonding geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N2—H2A...O2 ⁱ	0.86	2.15	2.995 (5)	168
N2—H2B...O1 ⁱⁱ	0.86	2.23	3.038 (5)	156
N2—H2B...N1	0.86	2.39	2.771 (5)	107
C6—H6...O1	0.98	2.39	2.840 (5)	107
C8—H8C...N1	0.96	2.59	2.919 (6)	100

Symmetry codes: (i) $2 - x, \frac{1}{2} + y, 1 - z$; (ii) $2 - x, \frac{1}{2} + y, 2 - z$.

H atoms were positioned geometrically (N—H = 0.86 Å and C—H = 0.96–0.98 Å) and refined using a riding model, with U_{iso} values constrained to be $1.5U_{\text{eq}}$ of the carrier atom for methyl H atoms and $1.2U_{\text{eq}}$ for the remaining H atoms. The Flack (1983) parameter could not be determined reliably because of insufficient anomalous scattering effects. During the final cycles of refinement, the Friedel pairs in the data set were merged.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

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